



TILMedia Suite 3

TLK-Thermo GmbH

in cooperation with

Institut für Thermodynamik

Technische Universität Braunschweig



TIL Media Substance properties optimized for stable and extremely fast dynamic simulations

- Calculation methods to express thermophysical properties of:



Incompressible Liquids



Ideal Gases



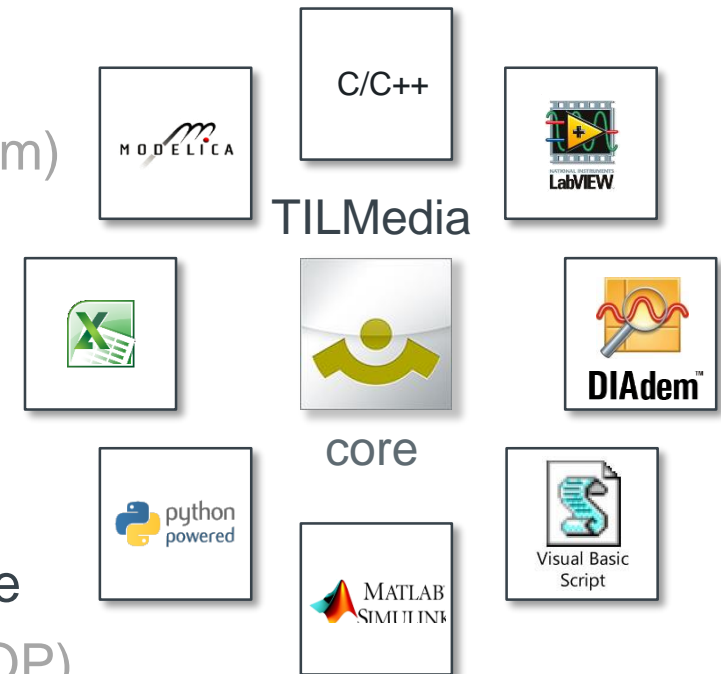
Real Fluids (with vapor liquid equilibrium)

- Mixtures

- Optimized mathematical equations with extremely high calculation speeds and high accuracies

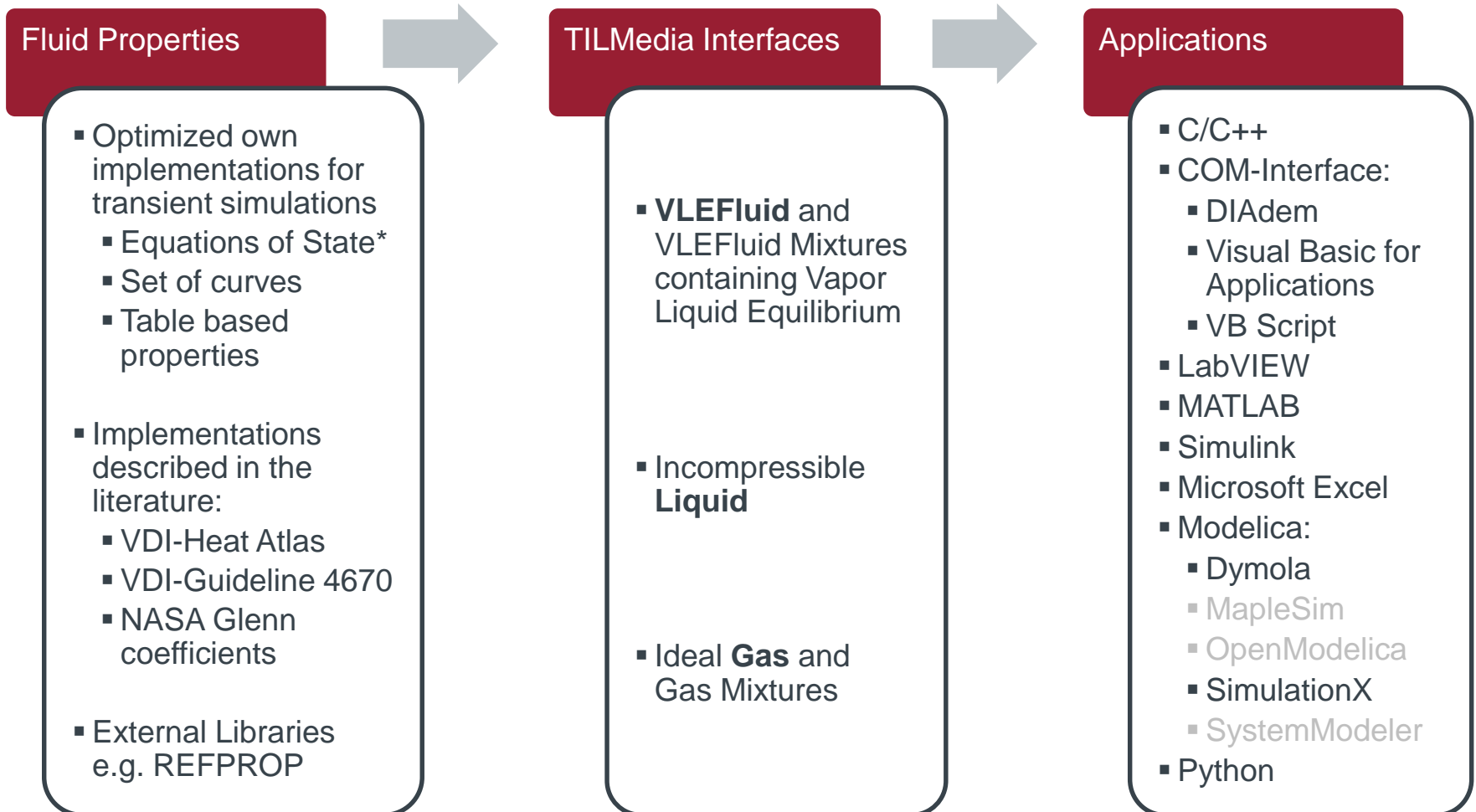
- Several hundreds of substances available (also from external sources e.g. REFPROP)

- TILMedia Suite interfaces one property core for various software





TILMedia as Interface



* In the following = EOS

▪ Not implemented jet



VLEFluids Available thermophysical properties

Interface for fluids containing a Vapor Liquid Equilibrium

The following VLEFluid properties are provided:

- density
- specific enthalpy
- pressure
- specific entropy
- temperature
- mass fraction
- mole fraction
- average molar mass
- steam mass fraction (quality)
- specific isobaric/isochoric heat capacity
- isobaric thermal expansion coefficient
- isothermal compressibility
- speed of sound
- derivative of density with respect to specific enthalpy
- derivative of density with respect to pressure
- derivative of density with respect to mass fraction
- heat capacity ratio / isentropic expansion factor
- molar mass of one component in mixture
- prandtl number
- thermal conductivity
- dynamic viscosity
- surface tension
- Saturation properties, critical properties & properties on dew and bubble line

As a function of:

- p, h, x_i
- p, T, x_i
- p, s, x_i
- d, T, x_i

with $x_i=1$ for a single component



Available pure VLEFluids – EOS

Approaches optimized for simulation with EOS

- faster than REFPROP
- not exportable
- moderate memory requirement
- very good accuracy of calculations
- Ammonia
- Argon
- CO₂ (Span/Wagner and GERG)
- Ethanol
- Ethylbenzene
- M-Xylene
- Nitrogen
- Oxygen
- O-Xylene
- Propane
- P-Xylene
- R1234yf
- R125
- R134a (Tillner-Roth/Baehr and Astina/Sato)
- R143a
- R245fa
- R32
- R404A (PPF)
- R407C (PPF)
- R410A (PPF)
- R507A (PPF)
- Water (IAPWS 1995)



Available pure VLEFluids

Implementations with sets of curves:

- very fast
- exportable
- low memory requirement
- good accuracy of calculations

- CO₂
- R1234yf
- R134a
- R407C
- R410A
- Water

Table based thermophysical properties*:

- very fast
- exportable
- high memory requirement
- very good accuracy of calculations

- Air (PPF)
- Methane
- R134a
- Table based properties of all REFPROP-Fluids possible on demand.

* Only available with the modelica library TIL



Available VLEFluids – REFPROP

REFPROP (208 media, pure fluids and mixtures):

▪ 1-Butene	▪ DEE	▪ Krypton	▪ Novec7000	▪ R-115	▪ R-407B	▪ R-423A	▪ R-508B
▪ Acetone	▪ Decane	▪ (R-784)	▪ Octane	▪ R-116	▪ R-407C	▪ R-424A	▪ R-509A
▪ Air	▪ DMC	▪ MDN	▪ Orthohydrogen	▪ R-123	▪ R-407D	▪ R-425A	▪ R-510A
▪ Amarillo	▪ DME	▪ MD2M	▪ (R-702)	▪ R-124	▪ R-407E	▪ R-426A	▪ R-512A
▪ Ammonia	▪ Ebenzene	▪ MD3N	▪ Oxygen	▪ R-125	▪ R-407F	▪ R-427A	▪ (Lemy 134)
▪ (R-717)	▪ Ekofisk	▪ MD4N	▪ (R-732)	▪ R-134a	▪ R-408A	▪ R-428A	▪ R-744
▪ Argon (R-740)	▪ Ethane	▪ Methane	▪ Oxylene	▪ R-141b	▪ R-409A	▪ R-429A	▪ R-1216
▪ Benzene	▪ (R-170)	▪ (R-50)	▪ Parahydrogen	▪ R-142b	▪ R-409B	▪ R-430A	▪ R-1233zd
▪ Butane	▪ Ethanol	▪ Methanol	▪ (R-702p)	▪ R-143a	▪ R-410A	▪ R-431A	▪ R-1234yf
▪ (R-600)	▪ Ethylene	▪ Methyl	▪ Pentane	▪ R-152a	▪ R-410B	▪ R-432A	▪ R-1234ze
▪ C1-CC6	▪ (R-1150)	▪ Linoleate	▪ (R-601)	▪ R-161	▪ R-411A	▪ R-433A	▪ R-C318
▪ C2-Butene	▪ Fluorine	▪ Methyl	▪ Propane	▪ R-218	▪ R-411B	▪ R-434A	▪ RE143a
▪ C3-CC6	▪ Gifcoast	▪ Linolenate	▪ (R-290)	▪ R-227ea	▪ R-412A	▪ R-435A	▪ (HFE-143m)
▪ C4-F10	▪ H2S	▪ MM	▪ Propylene	▪ R-236ea	▪ R-413A	▪ R-436A	▪ RE245cb2
▪ (R-3-1-10)	▪ HCL	▪ Methyl Oleate	▪ (R-1270)	▪ R-236fa	▪ R-414A	▪ R-436B	▪ (HFE-245cb2)
▪ C5-F12	▪ Helium	▪ Methyl	▪ Propyne	▪ R-245ca	▪ R-414B	▪ R-437A	▪ RE245fa2
▪ (R-4-1-12)	▪ (R-704)	▪ Palmitate	▪ P-Xylene	▪ R-245fa	▪ R-415A	▪ R-438A	▪ (HFE-245fa2)
▪ C11	▪ Heptane	▪ Methyl	▪ R-11	▪ R-365mfc	▪ R-415B	▪ R-441A	▪ RE347mcc
▪ C12	▪ Hexane	▪ Stearate	▪ R-12	▪ R-401A	▪ R-416A	▪ R-442A	▪ (HFE-7000)
▪ C-F3-I	▪ HighCO2	▪ M-Xylene	▪ R-13	▪ R-401B	▪ R-417A	▪ (RS-50)	▪ SF6
▪ CO	▪ HighN2	▪ N2O (R-744A)	▪ R-14	▪ R-401C	▪ R-418A	▪ R-443A	▪ SO2 (R-764)
▪ COS	▪ Hydrogen	▪ Neon (R-720)	▪ R-21	▪ R-402A	▪ R-419A	▪ R-444A (AC5)	▪ T2-Butene
▪ Cyclohexane	▪ (R-702)	▪ Neopentane	▪ R-22	▪ R-402B	▪ R-420A	▪ R-500	▪ Toluene
▪ Cyclopropane	▪ I-Butene	▪ NF3	▪ R-23	▪ R-403A	▪ R-421A	▪ R-501	▪ Water
▪ D2	▪ I-Hexane	▪ Ngsample	▪ R-32	▪ R-403B	▪ R-421B	▪ R-502	▪ Xenon
▪ D4	▪ I-Octane	▪ Nitrogen	▪ R-40	▪ R-404A	▪ R-422A	▪ R-503	
▪ D5	▪ I-Pentane	▪ (R-728)	▪ R-41	▪ R-405A	▪ R-422B	▪ R-504	
▪ D6	▪ (R-601a)	▪ Nonane	▪ R-113	▪ R-406A	▪ R-422C	▪ R-507A	
▪ D2O	▪ Isobutanol	▪ Novec649	▪ R-114	▪ R-407A	▪ R-422D	▪ R-508A	

Table based properties (high speed) can be created of all 208 REFPROP-Fluids on demand.



VLEFluid mixtures

Variable mixture calculations:

Fundamental EOS (very detailed, designed for transient simulations):

- Ammonia and Water (Tillner-Roth & Friend)

Cubic EOS:

- Argon
- CO₂
- Ethanol
- Hydrogen
- Nitrogen
- Oxygen
- R134a
- Water
- and many more (VDI-Wärmeatlas)

All 208 REFPROP-Fluids with each other



Liquids Available thermophysical properties

Interface for incompressible liquids

The following liquid properties are provided:

- density
- specific enthalpy
- pressure
- specific entropy
- temperature
- specific isobaric heat capacity
- isobaric thermal expansion coefficient
- derivative of density with respect to specific enthalpy
- prandtl number
- thermal conductivity
- dynamic viscosity

As a function of:

- p, h
- p, T



Available Liquids

Polynomial fits, 1-dimensional, temperature dependent:

- Addinol XW15
- Glysantin (30%-60%)
- Oil Aral 0W30
- Oil 15W40
- Propylenglykol (30%-50%)
- SHC_XMP320 (Syntetic gear oil)
- Therminol 59
- Therminol 66
- Therminol D12
- Tyfocor 30
- Tyfocor 45
- Tyfocorl 33
- Water
- Zitrec M10
- Zitrec M20

All liquid mediums listed in VDI-Heat Atlas



Gases Available thermophysical properties

Interface for gases and gas-vapor mixtures

The following gas properties are provided:

- density
- specific enthalpy
- pressure
- specific entropy
- temperature
- mass fraction
- mole fraction
- average molar mass
- derivative of density with respect to specific enthalpy
- derivative of density with respect to pressure
- derivative of density with respect to mass fraction
- partial pressure of components
- mass fraction of gasous condensing component
- specific isobaric/isochoric heat capacity
- isobaric thermal expansion coefficient
- isothermal compressibility
- speed of sound
- relative humidity
- prandtl number
- thermal conductivity
- dynamic viscosity
- Saturation properties

As a function of:

- p, h, xi
- p, T, xi
- p, s, xi

with $xi=1$ for a single component



Available pure gases

Approaches with EOS (high accuracy and fast):

- Dry Air
- Exhaust Gas
- Diesel Exhaust Gas

VDI-Guideline 4670:

- | | | |
|------------|---------|-------------------|
| ▪ Dry Air | ▪ Argon | ▪ Carbon Dioxide |
| ▪ Nitrogen | ▪ Neon | ▪ Carbon Monoxide |
| ▪ Oxygen | ▪ Water | ▪ Sulphur Dioxide |

All 275 mediums listed in VDI-Heat Atlas

All 2024 mediums listed by NASA Glenn coefficients



Gas mixtures

Variable gas mixture calculations with independent library choice (All pure gases can be mixed with each other):

- Approaches with EOS
- VDI-Guideline 4670
- VDI-Heat Atlas (275 media)
- NASA Glenn coefficients (2024 media)



Moist Air Available thermophysical properties

Interface for moist air (specialized form of gas mixture)

The following moist air properties are provided:

- density
- specific enthalpy
- pressure
- specific entropy
- temperature
- molar mass
- specific isobaric heat capacity
- specific isochoric heat capacity
- partial pressures
- (saturation) water mass fraction
- (saturation) water content
- (saturation) humidity ratio
- relative humidity
- specific enthalpy 1+x
- specific enthalpy of pure gas
- specific enthalpy of vaporisation
- specific enthalpy of desublimation
- prandtl number
- thermal conductivity
- dynamic viscosity
- freezing point
- speed of sound
- mass fraction
- gaseous mass fraction
- isobaric thermal expansion coefficient
- isothermal compressibility
- density derivative WRT mass fraction
- density derivative WRT pressure
- density derivative WRT specific enthalpy

As a function of:

- p, h, xi
- p, s, xi
- p, T, xi
- $p, T, \text{humRatio}$
- p, T, phi



Available Moist Air

Moist Air – TLK and IfT:

- Gas vapor mixture
- Heat capacity of water is considered as constant
- Condensing and ice build up realized by constant enthalpy of evaporation and constant enthalpy of fusion
- Transport properties equal to those of dry air

VDI-Guideline 4670 for Moist Air and combustion gases:

- Gas vapor mixture
- Condensing and ice build up realized by temperature dependent enthalpy of evaporation and temperature dependent enthalpy of fusion
- Transport properties equal to those of dry air



Example in Modelica

Input of e.g. medium name, enthalpy and pressure amongst others to calculate all thermophysical properties

General Advanced Add modifiers

Component

Name vleFluid1

Comment

Model

Path TILMedia.VLEFluid_ph

Comment VLE-Fluid model describing super-critical, subcooled, superheated fluid including the vapour liquid equilibrium (p, h and xi as independent variables)

Parameters

vleFluidType	TILMedia.CO2	type record of the VLE fluid or VLE fluid mixture
computeTransportProperties	false	=true, if transport properties are calculated
computeVLEAdditionalProperties	false	Compute detailed vapour liquid equilibrium properties
computeVLETransportProperties	false	Compute detailed vapour liquid equilibrium transport properties
deactivateTwoPhaseRegion	false	Deactivate calculation of two phase region
h	h J/kg	Specific enthalpy
p	p Pa	Pressure
xi	vleFluidType.xi_default	Mass Fraction of Component i

OK Info Cancel

Modeling Simulation



Example in MATLAB

Anwendungserweiterung

- TILMediaMatlab321x64.dll
- TILMediaMatlab321Win32.dll

C/C++ Header

- TILMediaHeader4Matlab.h
- portable.h

Class

- VLEFluid.m
- MoistAir.m
- Liquid.m
- Gas.m

Function

- loadTILMedia
- checkInputVar
- calcComputeF

```
% TILMedia-functions:
```

```
vle = VLEFluid();  
vle = vle.setVLEFluidType(vle_name,1);
```

```
% Calculation-loop for thermophysical properties:
```

```
for i=1:length(p_vle)  
    vle = vle.setState_phxi(p_vle(i)*ones(length(h_vle),1),h_vle);  
    T_vle = [T_vle,vle.T];  
    s_vle = [s_vle,vle.s];  
end
```

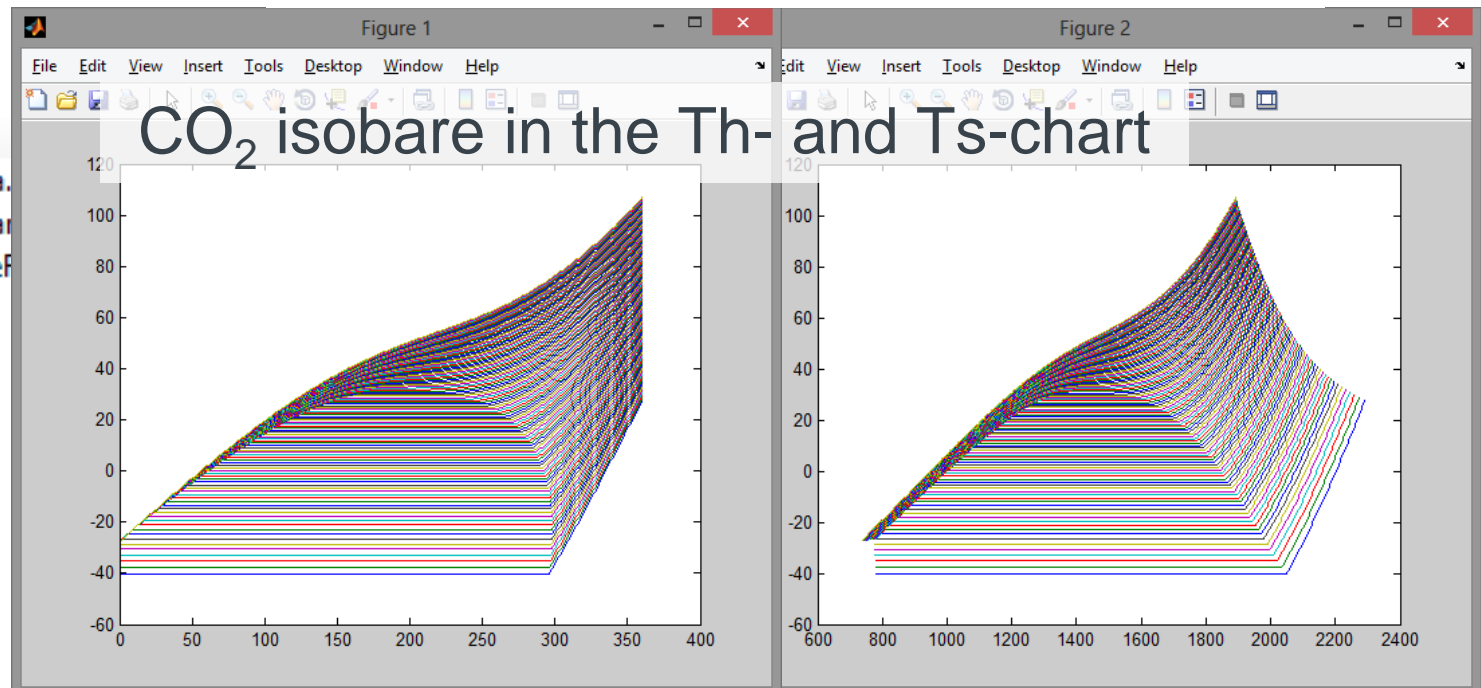
```
% Choise of substance:
```

```
vle_name = 'CO2';
```

```
% Input of enthalpy- and pressure-range:
```

```
h_vle = [140:1:500]*1e3;
```

```
p_vle = [10:1:120]*1e5;
```



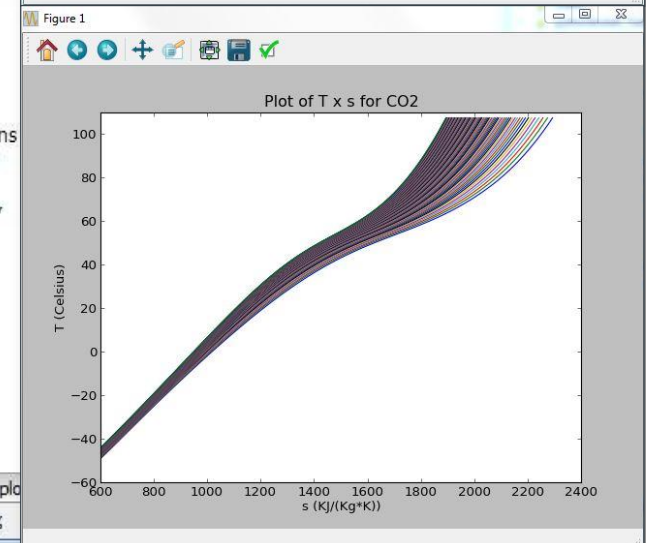
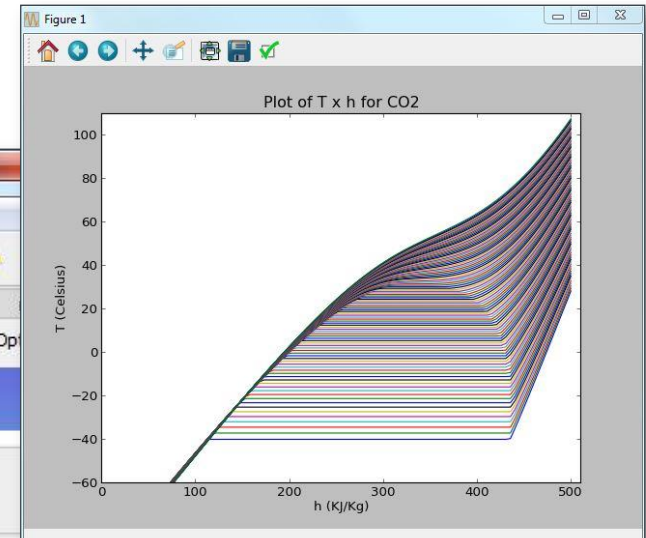
Example in Python

CO₂ isobare in the Th- and Ts-chart

```

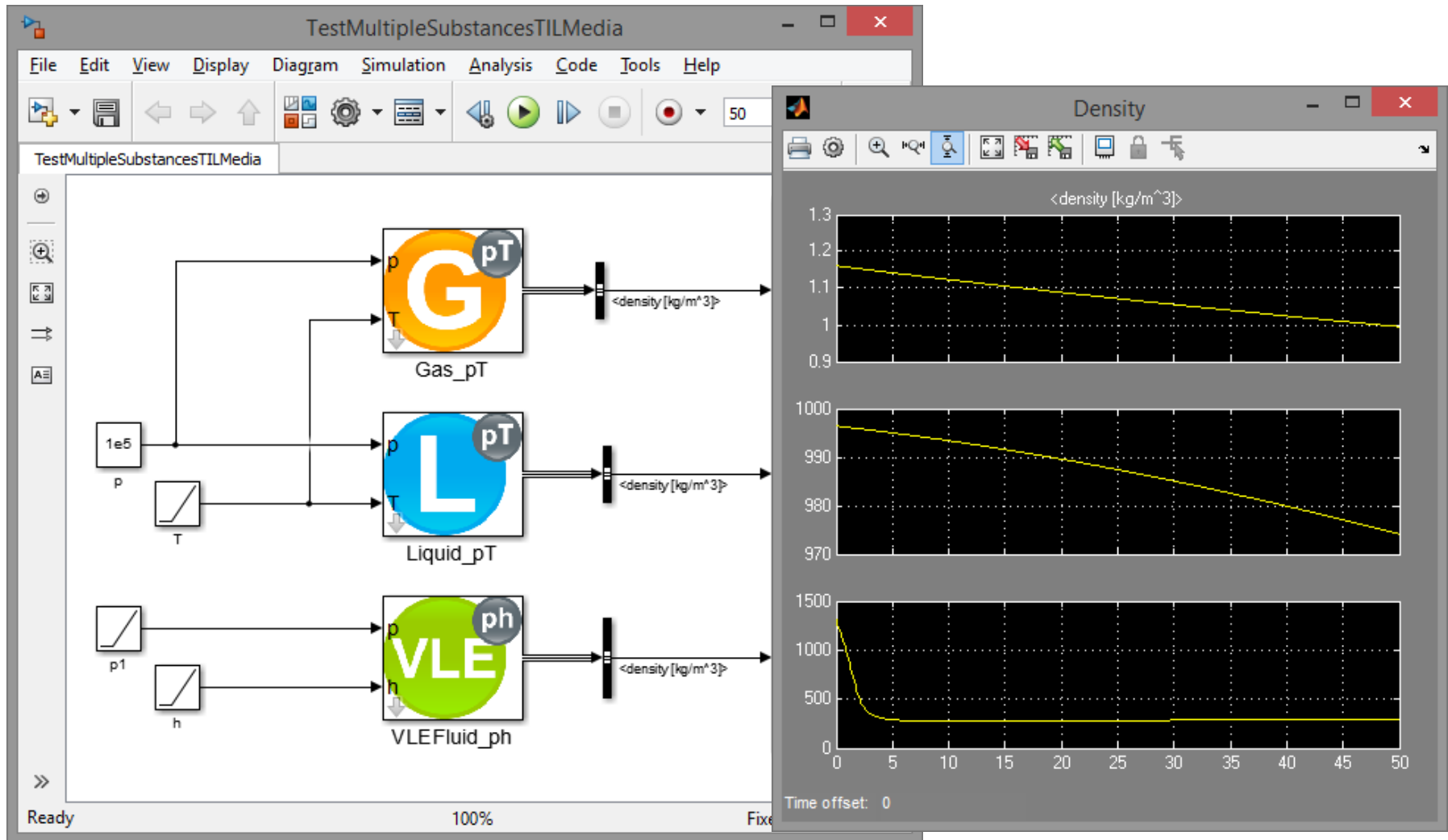
Spyder (Python 2.7)
File Edit Search Source Run Interpreters Tools View ?
Editor - F:\TILMedia\isobar_T_x_h.py
Object inspector
Source Console Object a Op
range
Definition : range(stop)
Type : Function of __builtin__ module
range(stop) -> list of integers
range(start, stop[, step]) -> list of integers
Return a list containing an arithmetic progression of integers. range(i, j) returns [i, i+1, i+2, ..., j-1]; start (!) defaults to 0. When step is given, it specifies the increment (or decrement). For example, range(4) returns [0, 1, 2, 3]. The end point is omitted! These are exactly the valid indices for a list of 4 elements.

1 #isobar.py
2 import TILMedia
3 import numpy as np
4 import pylab as pl
5
6 nrElem_p = 100
7 nrElem_h = 100
8
9 p = np.linspace(10,120,nrElem_p) # bar
10 h = np.linspace(10,500,nrElem_h) # KJ/Kg
11 T = np.zeros(nrElem_h) # Celsius
12 vle = TILMedia.VLEFluid('Refprop.CO2')
13
14 for i_p in range(nrElem_p):
15     for i_h in range(nrElem_h):
16         vle.setState_phxi(p[i_p]*1e5,h[i_h]*1000)
17         T[i_h] = vle.T - 273.15
18     pl.plot(h,T)
19
20 # give plot a title
21 pl.title('Plot of T x h for CO2')
22 # make axis labels
23 pl.xlabel('h (KJ/Kg)')
24 pl.ylabel('T (Celsius)')
25 # set axis limits
26 pl.xlim(0, 510)
27 pl.ylim(-60.0, 110.)
28 # show the plot on the screen
  
```



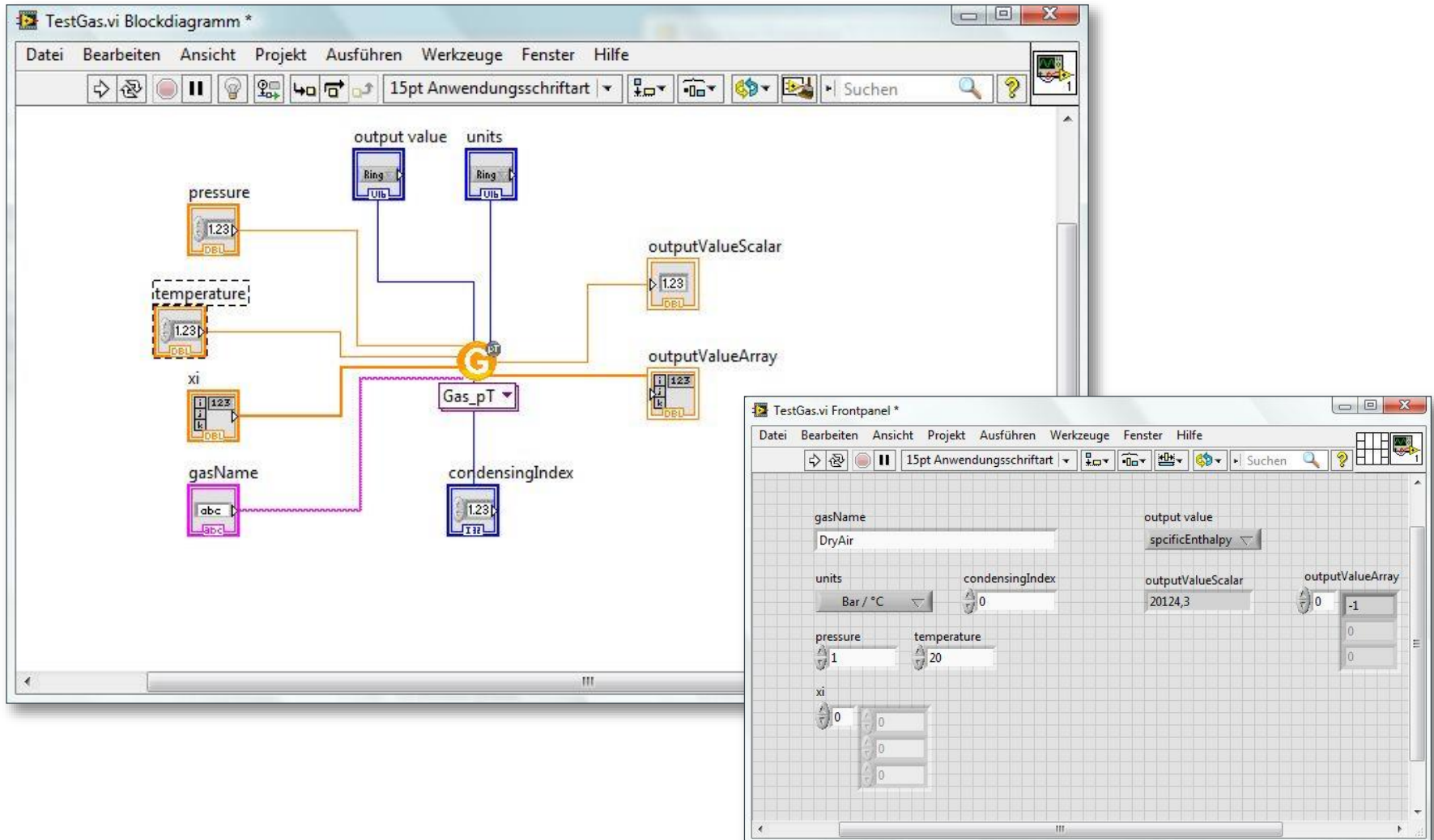


Example in Simulink





Example in LabVIEW





Example in Excel

BEREICHE				
=Liquid_density_T(A3;C5)				
	A	B	C	D
1	Description	Variable	Value	Unit
2	Inputs:			
3	Glystantin_50			
4	Pressure	p	1.01E+05	Pa
5	Temperature	T	300	K
6	Calculation:			
7	Density	rho	=Liquid_density_T(A3;C5)	
8	Entropy	s	-567.3796701	J/(kg K)
9	Enthalpy	h	89323.76775	J/kg
10	Specific isobaric heat capacity	cp		
11	Isobaric thermal expansion coefficient	beta		
12	Dynamic viscosity	eta		
13	Thermal conductivity	lambda		
14	Prandtl number	Pr		

Calculation of density of water-glycol 50:50 at 300 K (and normal pressure)

Funktionsargumente

Liquid_density_T

LiquidName = "Glystantin_50"

T = 300

= 1065.506241

'Density' in [kg/m^3] as a function of T.

LiquidName Liquid name.

Formelergbnis = 1065.506241

[Hilfe für diese Funktion](#)

OK Abbrechen



Examples using Windows COM-Interface

Visual Basic for Applications (VBA) and Visual Basic Script (VBS)

```
'creating two variables
Dim obj, msg As String
'creating a liquid-object
Set obj = CreateObject("TILMedia.Liquid")
'set medium to water-glycol-mixture 50:50
Call obj.setLiquidType("Glysantin_50")
'calculating properties with pressure = 1
    bar and temperature = 300 K
Call obj.setState_pTxi(1e5, 300)
'constructing a message:
msg = "The density of Glysantin_50 is "
    + str(obj.d)
msg = msg + " under the conditions
    pressure = " + str(obj.p)
msg = msg + " and temperature = "
    + str(obj.T)
MsgBox msg 'displaying a message
```

DIAdem

```
'loading a data file with values for pressure
    and temperature
Call DataFileLoad(CurrentScriptPath
    &"Example_pT.TDM", "TDM", "Load")
'creating four variables
Dim obj, array_p, array_T, chnName_d(0)
'creating a vector-liquid-object
Set obj = CreateObject("TILMedia.VectorLiquid")
'set medium to water-glycol-mixture 50:50
Call obj.setLiquidType("Glysantin_50")
'store pressure values of channel to array
array_p = ChnToValue("channelName_p")
'temperature values from channel to array
array_T = ChnToValue("channelName_T") Call
'calculating properties with pressure and
    temperature array in SI units
obj.setState_pTxi(array_p, array_T)
'set new channel name (vector) to "density"
chnName_d(0) = "density"
'saving density values in channel
Call ArrayToChannels(obj.d, chnName_d, true)
```

red colored = DIAdem-specific functions

Thank you



If you have any questions,
don't hesitate to contact us at
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